

# Radioactive Decay Study Guide



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Define the following terms: Stable- Not likely to change or fail; firmly established Unstable- Prone to change, fail, or give way; not stable Radioactive Decay- The

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21.1 & 21.2 Radioactive Decay. Study Guide. Directions: Fill in the blank with the correct term. The number 238 in Uranium-238 represents the \_\_\_\_\_. The

## **Section 4.4: unstable nuclei and radioactive decay**

Students are to complete their Science Notebook reading Guide pages 54 and 55 on unstable nuclei and radioactive decay. We will answer these two pages and reinforce

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Explains radioactive decay, including parent and daughter isotopes, and half lives, and its importance.

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43 terms In nuclear chemistry, an atom is referred to as a nuclide, What does the 4 in  $4\text{He}^2$  represent? the mass number, What does the 101 in  $^{256}\text{Md}_{101}$

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AP Chemistry Study Guide: Chapter 19, Nuclear Products from a nuclear fission of a uranium atom such as  $^{90}\text{Sr}$  and  $^{137}\text{Cs}$  are highly radioactive and decay by

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